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chapter22 study guide for content pdf - In chemistry, pH (/ p iĒ• Ē^ eĒ^a tĒf /) is a logarithmic scale used to specify the acidity or basicity of an aqueous solution. It is approximately the negative of the base 10 logarithm of the molar concentration, measured in units of moles per liter, of hydrogen ions. More precisely it is the negative of the base 10 logarithm of the activity of the hydrogen ion. ... pH - Wikipedia -

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